

Name:

Date:

Section:

Water Potential Activity: Water Movement in Plants

Cells and Water: A Balancing Act! 💧

Phase 1: ENGAGE (5 minutes)

- *Getting Started:**

Open peebedu.com and navigate to Interactive Water Potential Simulation

- *First Impressions:**

1. Look at the beaker - what do you see?

- Controls on right = _____

1. Click "Add Solute" - watch what happens!

1. **Big Question:** 🤔

Why do your fingers get wrinkly in the bath but not in the rain?

- *Fun Fact:** Your body has about 37 trillion cells, each balancing water! 💧


Phase 2: EXPLORE (18 minutes)

- *Adventure 1: Making Solutions** 

Let's add different "seasonings" to our water!

- *Solution Recipe Book.**

What I Added	How Many Clicks	What Happened to Numbers
Sugar (sucrose)	5	Water potential = _____
Salt (NaCl)	5	Water potential = _____
Road salt (CaCl ₂)	5	Water potential = _____


- *Discovery.** Which one changed the water potential the MOST? _____
- *Adventure 2: Cell Testing Lab** 

Time to add cells! Try each one in plain water first:

- *Cell Observation Journal.**

Draw what happens to each cell:

Cell Type	In Plain Water	Draw It!
Plant cell	Swells / Shrinks / Same	[Box]
Animal cell	Swells / Shrinks / Same	[Box]
Red blood cell	Swells / Shrinks / Same	[Box]

- *Think.** Why don't they all do the same thing? _____
- *Adventure 3: The Goldilocks Challenge** 

Find the "just right" solution for each cell (where it stays the same size):

- *Perfect Match Game.**

Cell Type	Solute I Used	How Much?	Success?
Red blood cell	✓ or ✗		
Plant cell	✓ or ✗		
Bacteria	✓ or ✗		

- *Pattern Alert:** Which cell needs the MOST salty water? _____
- *Adventure 4: Extreme Conditions!** 🌡️

Pick a plant cell and put it in salty water. Now change things:

- *Environmental Testing:**

What I Did	What Happened to Cell	Why I Think This Happened
Made it hot (50°C)		
Made it cold (0°C)		
Evaporated water		
Added more water		

Phase 3: EXPLAIN (12 minutes)

- *How Water Moves** 💧 ➡️

1. Water's Rules:

Water is like kids at recess - it goes where there's more room!

Fill in the rule:

...

Water moves FROM: _____ water potential (less stuff dissolved)

Water moves TO: _____ water potential (more stuff dissolved)

...

1. The Magic Equation (simplified):

Lower water potential = More dissolved stuff = Water wants to go there!

Match these:

...

Pure water • • Very negative number

A little salt water • • Zero

Very salty water • • Small negative number

...

1. Cell Survival Strategies:

How do cells deal with water?

- **Too much water:** Cell might _____

- **Just right:** Cell stays _____

1. Real-Life Examples:

Draw lines to match:

...

Pickle in salt water • • Cells swell up

Raisin in water • • Cells shrink

Fresh vegetables • • Cells stay firm

Slug with salt • • Cells lose water fast

...

Phase 4: ELABORATE (8 minutes)

- *Water Potential in Your World!** 🌍

1. Kitchen Science:

Why do we put salt on icy roads?

- Salt makes water potential: Higher / Lower (circle one)

- Cells nearby might:

1. Garden Mystery:

Your mom waters the plants, but accidentally uses salt water!

- What happens to plant cells?

- How to save the plants?

1. Ocean Life:

How do fish survive in the salty ocean?

Problem	Fish Solution
Ocean is VERY salty	
Cells could shrink	

1. Sports Drinks:

Why do athletes drink special drinks instead of just water?

Phase 5: EVALUATE (7 minutes)

- *Show What You Know!** 🎯

1. Fill in the Blanks:

When you put a cell in _____ water, water moves into the cell.

When you put a cell in _____ water, water moves out of the cell.

The cell stays the same in _____ solutions.

1. True or False: (Circle your answer)

- T / F Water always moves from high to low water potential
- T / F All cells react the same way to salt water
- T / F Temperature doesn't affect water movement
- T / F Plant cells can burst from too much water
- T / F Your blood is isotonic to your cells

1. Drawing Challenge:

Draw what happens to a red blood cell in:

...

Pure Water: Salt Water: Blood:

[Box] [Box] [Box]

...

1. Problem Solving: 🧩

Your goldfish bowl accidentally gets salt in it!

- Fish cells normally have $\Psi = -0.5$
- Salt water now has $\Psi = -2.0$

- What should you do?

1. Create a Warning Label:

Make a warning label for salt near a garden:

⚠ WARNING:

- *Exit Ticket:** 

Rate your understanding:

- Water movement: 😐 😐 😊 😊
- Why cells shrink/swell: 😐 😐 😊 😊
- Using the simulation: 😐 😐 😊 😊

One cool thing I learned:

One question I still have:

- --

Fun Extensions:

1. Home Experiment:

- Put gummy bears in different liquids
- Measure size after 24 hours
- Graph your results

- Explain using water potential

1. **Cell Comic Strip:**

- Create a 6-panel comic
- Show a cell's adventure
- Include different solutions
- Add speech bubbles!

1. **Design a Cell Spa:**

- What solutions would you offer?
- Create a menu for different cells
- Include "treatments" and effects
- Make it funny but scientific!

****Amazing Facts!**** ✨

- A camel's red blood cells are oval to survive dehydration!
- Plants can create 300 pounds of pressure with water
- Your kidneys filter 180 liters of fluid daily
- Jellyfish are 95% water
- Seeds can survive with almost no water for years!

Key Vocabulary:

See activity for vocabulary specific to this topic.